

○ Non ideal classical gas

monatomic gas (N identical particles of mass m) in volume V
at temperature T

$\left. \begin{array}{l} T \text{ is sufficiently high} \\ n = N/V \sim \text{sufficiently low} \end{array} \right\} \text{classical limit}$

$$H = K + U$$

$$K = \frac{1}{2m} \sum_{j=1}^N \vec{p}_j^2 \quad \text{kinetic}$$

$$U = \frac{1}{2} \sum_{j=1}^N \sum_{k=1, k \neq j}^N u_{jk} \quad \text{interaction between particles}$$

$u_{jk} \rightarrow$ semiempirical potential "Lennard-Jones potential"

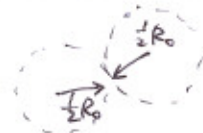
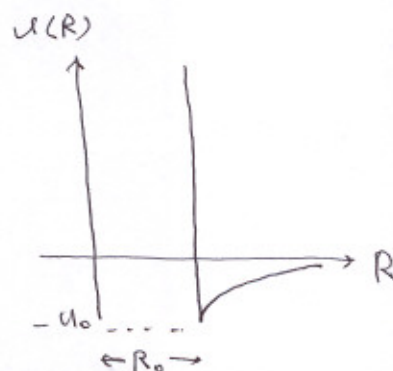
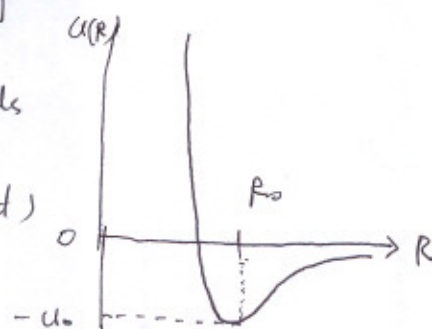
$$u(R) = u_0 \left[\underbrace{\left(\frac{R_0}{R} \right)^{12}}_{\text{exclusion}} - 2 \underbrace{\left(\frac{R_0}{R} \right)^6}_{\text{van der Waals attraction (dipole moment)}} \right]$$

further simplification

$$u(R) = \begin{cases} \infty & R < R_0 \\ -u_0 \left(\frac{R_0}{R} \right)^6 & R > R_0 \end{cases}$$

$R < R_0$

$R > R_0$



"weakly interacting hard sphere with radius $\frac{1}{2}R_0$ "

$$Z = \frac{1}{N!} \int \frac{1}{h^{3N}} \prod_{i=1}^N d^3\vec{r}_i d^3\vec{p}_i e^{-\beta(K+U)}$$

$$= \frac{1}{N!} \left(\frac{2\pi m}{h^2 \beta} \right)^{\frac{3}{2}N} Z_U, \quad Z_U = \int d^3\vec{r}_1 \dots d^3\vec{r}_N e^{-\beta U(\vec{r}_1, \vec{r}_2, \dots, \vec{r}_N)}$$

note: $\beta \rightarrow 0$ (high temperature) $\Rightarrow Z_U \rightarrow V^N$

($\because e^{-\beta U} \rightarrow 1$)

= ideal gas case

mean potential energy

$$\bar{U} = \frac{1}{2} N(N-1) \bar{u} \approx \frac{1}{2} N^2 \bar{u}$$

$$\bar{u} = \bar{u}_{jk} = \frac{\int e^{-\beta u_{jk}(\vec{r}_{jk})} u_{jk}(\vec{r}_{jk}) d^3\vec{r}_{jk}}{\int e^{-\beta u_{jk}(\vec{r}_{jk})} d^3\vec{r}_{jk}} \quad \left(\begin{array}{l} \vec{r}_{jk} = \vec{r}_j - \vec{r}_k \\ u_{jk}(\vec{r}_{jk}) = u_{jk}(\vec{r}_{jk}) \end{array} \right)$$

$$= - \frac{\partial}{\partial \beta} \ln \int e^{-\beta u(r)} d^3r \quad \downarrow \text{drop index } j, k$$

$u(r) \sim \text{small}$ except for small r

$$\Rightarrow e^{-\beta u(r)} \approx 1$$

$$\Rightarrow \int e^{-\beta u(r)} d^3r = \int (1 + (e^{-\beta u} - 1)) d^3r = V + I(\beta)$$

$$I(\beta) = \int (e^{-\beta u} - 1) d^3r = \int_0^\infty 4\pi r^2 (e^{-\beta u} - 1) dr$$

$$\Rightarrow \bar{u} = - \frac{\partial}{\partial \beta} \left[\ln V + \ln \left(1 + \frac{I}{V} \right) \right]$$

$$\approx - \frac{\partial}{\partial \beta} \frac{I}{V}$$

$$\downarrow \quad \frac{I}{V} \ll 1$$

$$\Rightarrow \bar{U} = - \frac{1}{2} \frac{N^2}{V} \cdot \frac{\partial I}{\partial \beta}$$

$$\bar{U} = - \frac{\partial}{\partial \beta} \ln Z_U \Rightarrow \ln Z_U(\beta) = \ln Z_U(\beta=0) - \int_0^\beta \bar{U}(\beta') d\beta' \quad \left. \begin{array}{l} Z_U(\beta=0) \\ = V^N \end{array} \right\}$$

$$= N \ln V + \frac{1}{2} \frac{N^2}{V} I(\beta)$$

- Equation of state, virial expansion

$$\bar{p} = \frac{1}{\beta} \frac{\partial \ln Z}{\partial V} = \frac{1}{\beta} \frac{\partial \ln Z_u}{\partial V}$$

$$\Rightarrow \beta \bar{p} = \frac{N}{V} - \frac{1}{2} \frac{N^2}{V^2} \bar{I}$$

$$\text{X: } \beta \bar{p} = \frac{\bar{p}}{kT} = n + B_2(T) n^2 + B_3(T) n^3 + \dots$$

$$n = \frac{N}{V}$$

$B_i \sim$ virial coefficient

"virial expansion"

(remember $n \rightarrow$ very small in classical limit)

$$B_2 = -\frac{1}{2} \bar{I}$$

$$= -2\pi \int_0^\infty (e^{-\beta u(R)} - 1) R^2 dR$$

$$\text{Use } u(R) = \begin{cases} \infty & (R < R_0) \\ -u_0 \left(\frac{R_0}{R}\right)^s & (R > R_0) \end{cases}$$

$$\Rightarrow B_2 = 2\pi \int_0^{R_0} R^2 dR - 2\pi \int_{R_0}^\infty (e^{-\beta u} - 1) R^2 dR$$

assume high temperature $\beta u_0 \ll 1$

$$\Rightarrow B_2 = \frac{2\pi}{3} R_0^3 - 2\pi \beta u_0 \int_{R_0}^\infty \left(\frac{R_0}{R}\right)^s R^2 dR$$

$$= \frac{2\pi}{3} R_0^3 \left(1 - \frac{3}{s-3} \frac{u_0}{kT}\right)$$

↓ assume $s > 3$

$$= b' - \frac{a'}{kT}$$

$$a' \propto u_0 R_0^3$$

$$\Rightarrow \frac{\bar{p}}{kT} = n + \left(b' - \frac{a'}{kT}\right) n^2$$

assume:
 $b'n \ll 1$ (dilute)

$$\Rightarrow \bar{p} + a'n^2 = nkT(1 + b'n) \simeq \frac{nkT}{1 - b'n}$$

$$\Rightarrow (\bar{p} + a'n^2) \left(\frac{1}{n} - b'\right) = kT \rightarrow \text{"van der Waals eq."}$$

$$n = \frac{N}{V} = \frac{N_A}{v} \leftarrow \text{Avogadro's \#}$$

(X: see also Reif section 10.5)

$$\Rightarrow \left(\bar{p} + \frac{a}{v^2}\right)(v - b) = RT$$